

ERRATA

SOLUTIONS MANUAL TO MOLECULAR THERMODYNAMICS OF FLUID-PHASE EQUILIBRIA

3rd EDITION (PRENTICE-HALL, 1999)

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Page	Location	Where is	Should be
46	figure	$(c - c_0), \text{g L}^{-1}$	$\pi(c - c_0), \text{Pa L g}^{-1}$
46	figure	$\pi(c - c_0), \text{Pa L g}^{-1}$	$(c - c_0), \text{g L}^{-1}$
128	Eq. (1)	$J_i = \frac{D_i}{\delta_M} S_{iF}^G P_{iF} - S_{iP}^G P_{iP}^G$	$J_i = \frac{D_i}{\delta_M} (S_{iF}^G P_{iF} - S_{iP}^G P_{iP}^G)$
129	Eq. (1)	$J_1 = \frac{D_1}{\delta_M} c_{1F}^M - c_{1P}^M$	$J_1 = \frac{D_1}{\delta_M} (c_{1F}^M - c_{1P}^M)$
130	Eq. (7)	$J_1 \left(\frac{\text{mol}}{\text{cm}^3 \text{ s}} \right)$	$J_1 \left(\frac{\text{mol}}{\text{cm}^2 \text{ s}} \right)$
130	Eq. (8)	$J_2 \left(\frac{\text{mol}}{\text{cm}^3 \text{ s}} \right)$	$J_2 \left(\frac{\text{mol}}{\text{cm}^2 \text{ s}} \right)$